

The background of the slide is a grayscale image of the moon's surface, showing various craters and lunar maria. A white rectangular box with a black border is centered on the slide, containing the title and author information.

# **Enceladus: A Habitable Environment?**

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# Hydrogen: So what...it's everywhere!

## Solar Abundances

Element	$N(\text{El})_0$
H .....	$2.431 \times 10^{10}$
He .....	$2.343 \times 10^9$
Li .....	55.47
Be .....	0.7374
B .....	17.32
C .....	$7.079 \times 10^6$
N .....	$1.950 \times 10^6$
O .....	$1.413 \times 10^7$
F .....	841.1
Ne .....	$2.148 \times 10^6$
Na .....	$5.751 \times 10^4$
Mg .....	$1.020 \times 10^6$
Al .....	$8.410 \times 10^4$
Si .....	$\equiv 1.00 \times 10^6$
P .....	8373
S .....	$4.449 \times 10^5$
Cl .....	5237
Ar .....	$1.025 \times 10^5$

Lodders (2003)



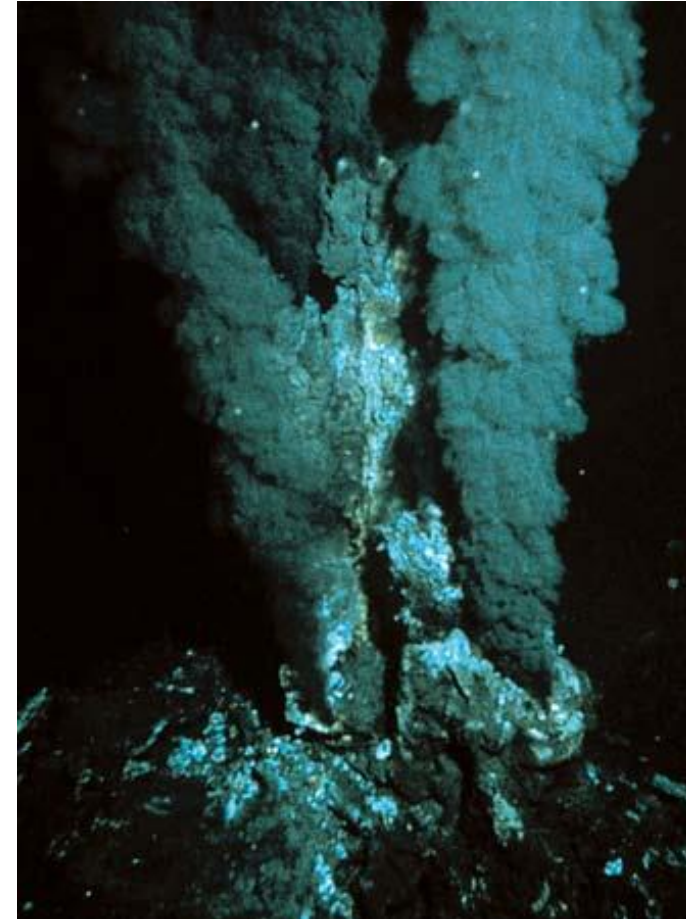


# But, H<sub>2</sub> is relatively rare on Ocean Worlds

0.55 ppm H<sub>2</sub> in Earth's atmosphere

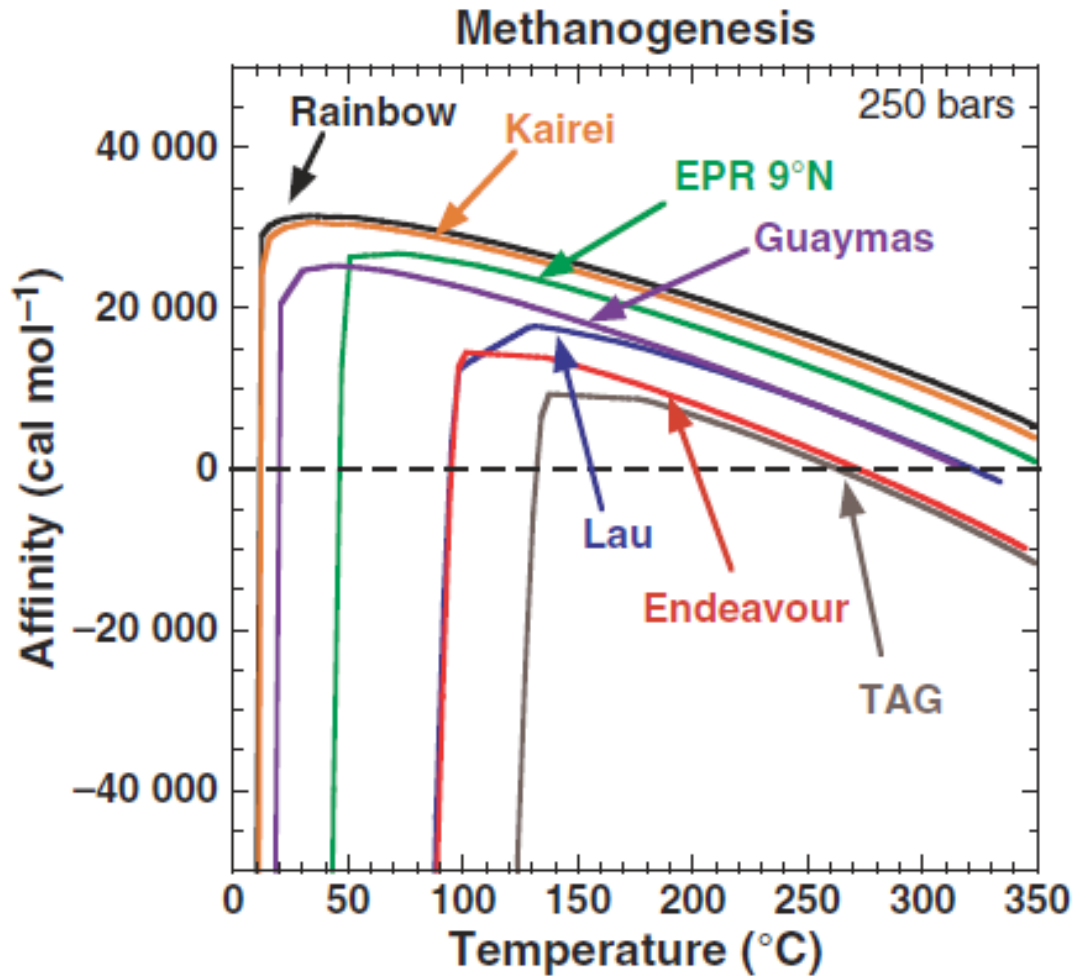


It's hard to hang on to H<sub>2</sub>

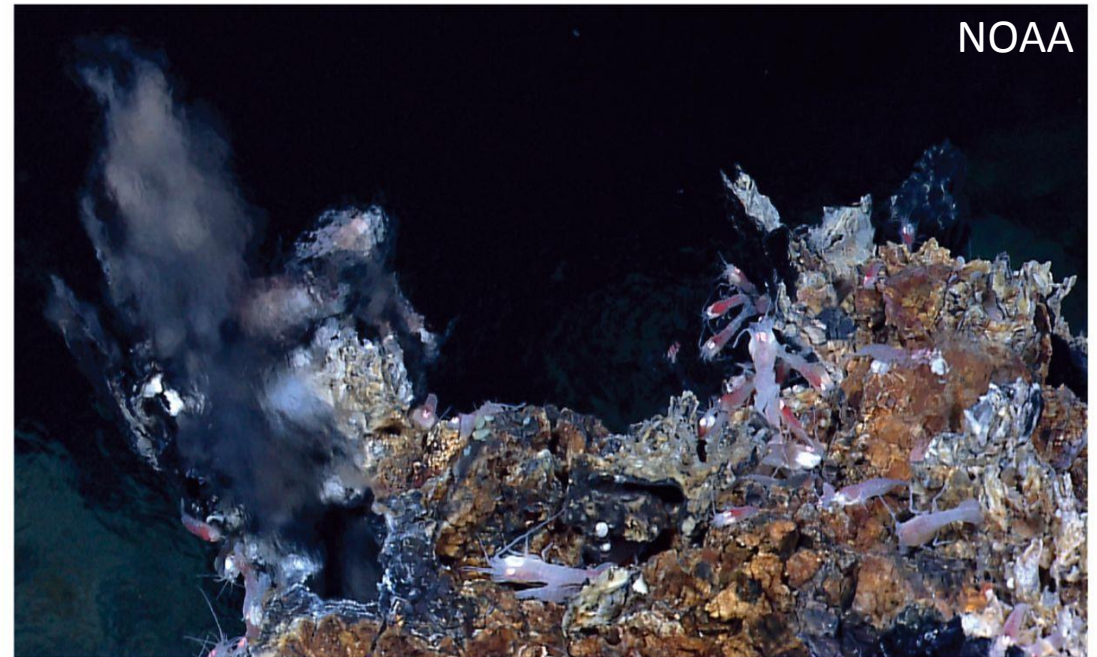
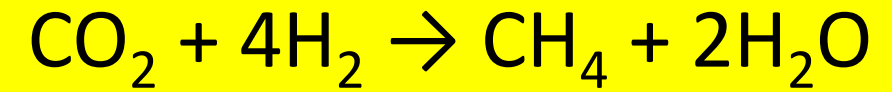


Water-rock hydrothermal processes produce of order 1 M tonnes H<sub>2</sub> per year (Sherwood Lollar et al., 2014)

# H<sub>2</sub> on Ocean Worlds: An Energy Source for Life



Shock & Canovas (2010)

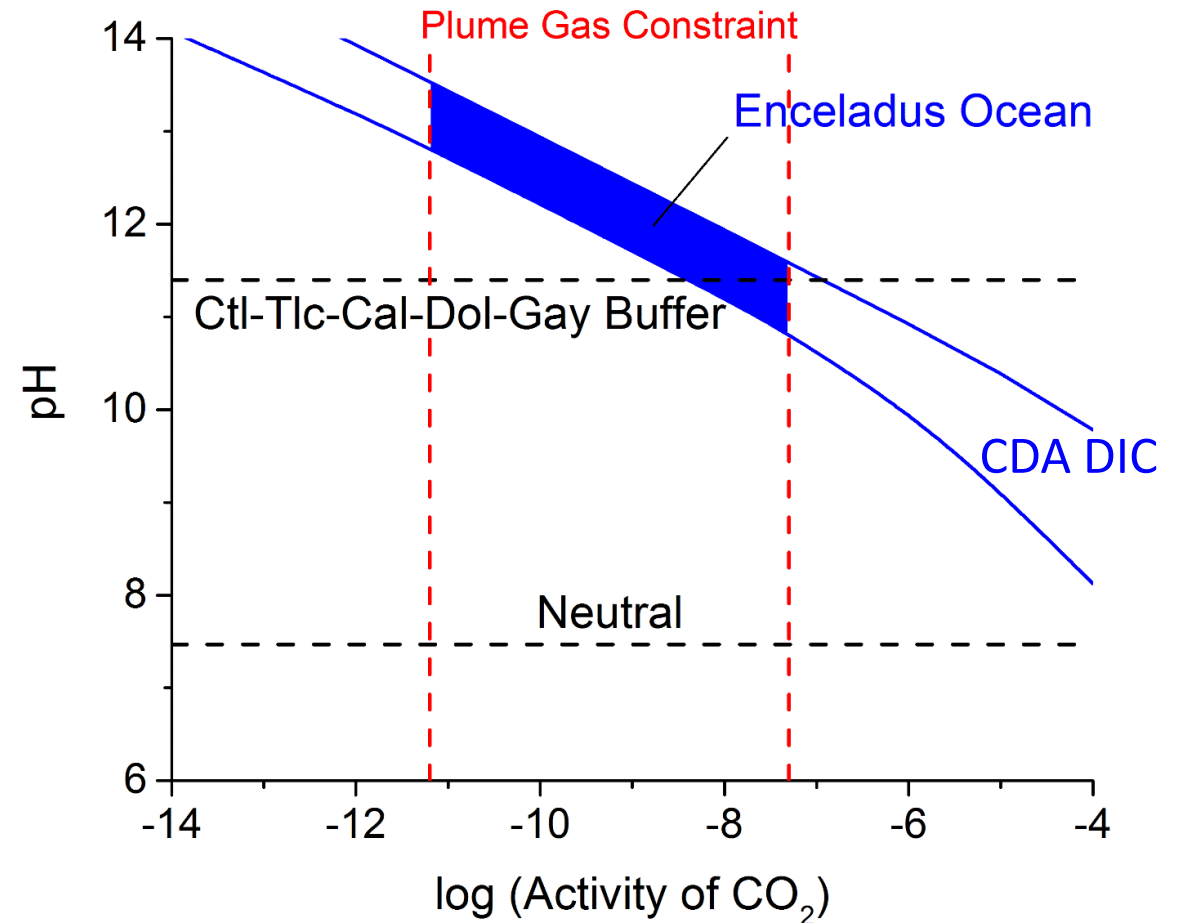


Chemical energy → Ecosystem

# Hydrothermal ultramafic rock alteration (serpentinization) on Enceladus



**Silica particles** (Hsu et al., 2015)  
form when hot vent fluids mix with  
cold ocean water at the ocean floor



**Alkaline ocean** (Glein et al., 2015)



# What exactly is serpentinization?

A Geochemical Process



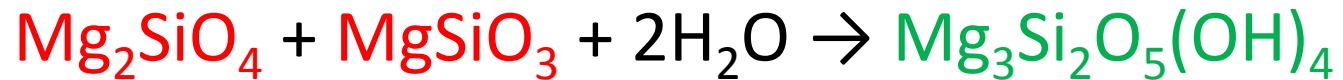
No!



California  
state rock

Yes!

For example:



Ultramafic Rock

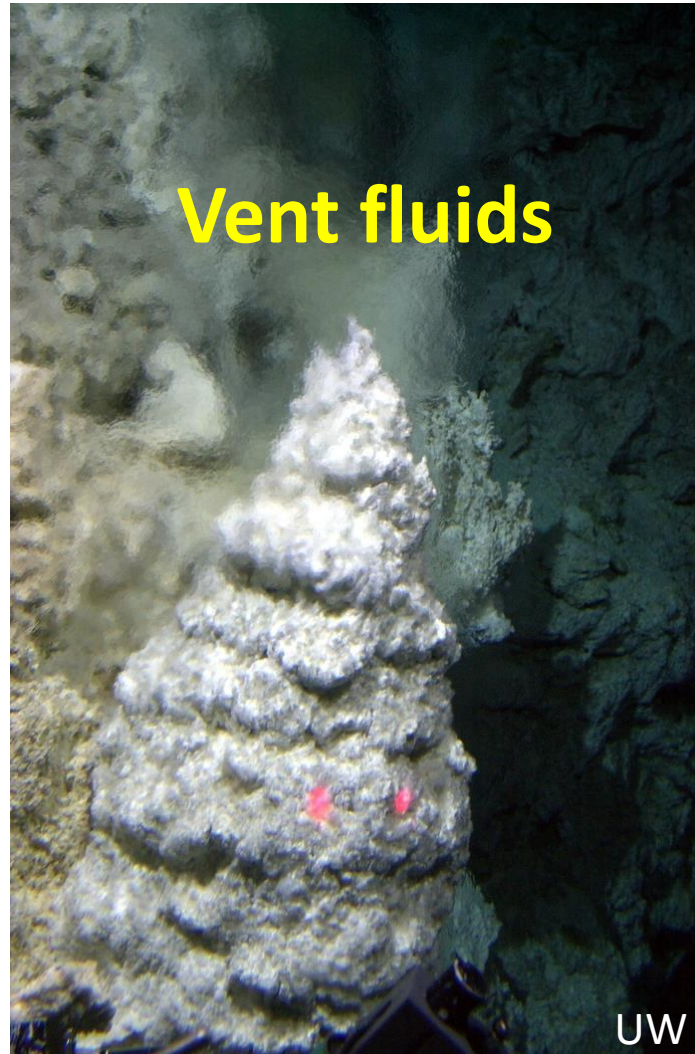
Serpentine (aka asbestos)



Serpentinization leads to high pH (rock enriched in bases)  
and...

# Serpentinizing hydrothermal systems on Earth produce large quantities of H<sub>2</sub>

## Lost City as a geochemical analogue of Enceladus

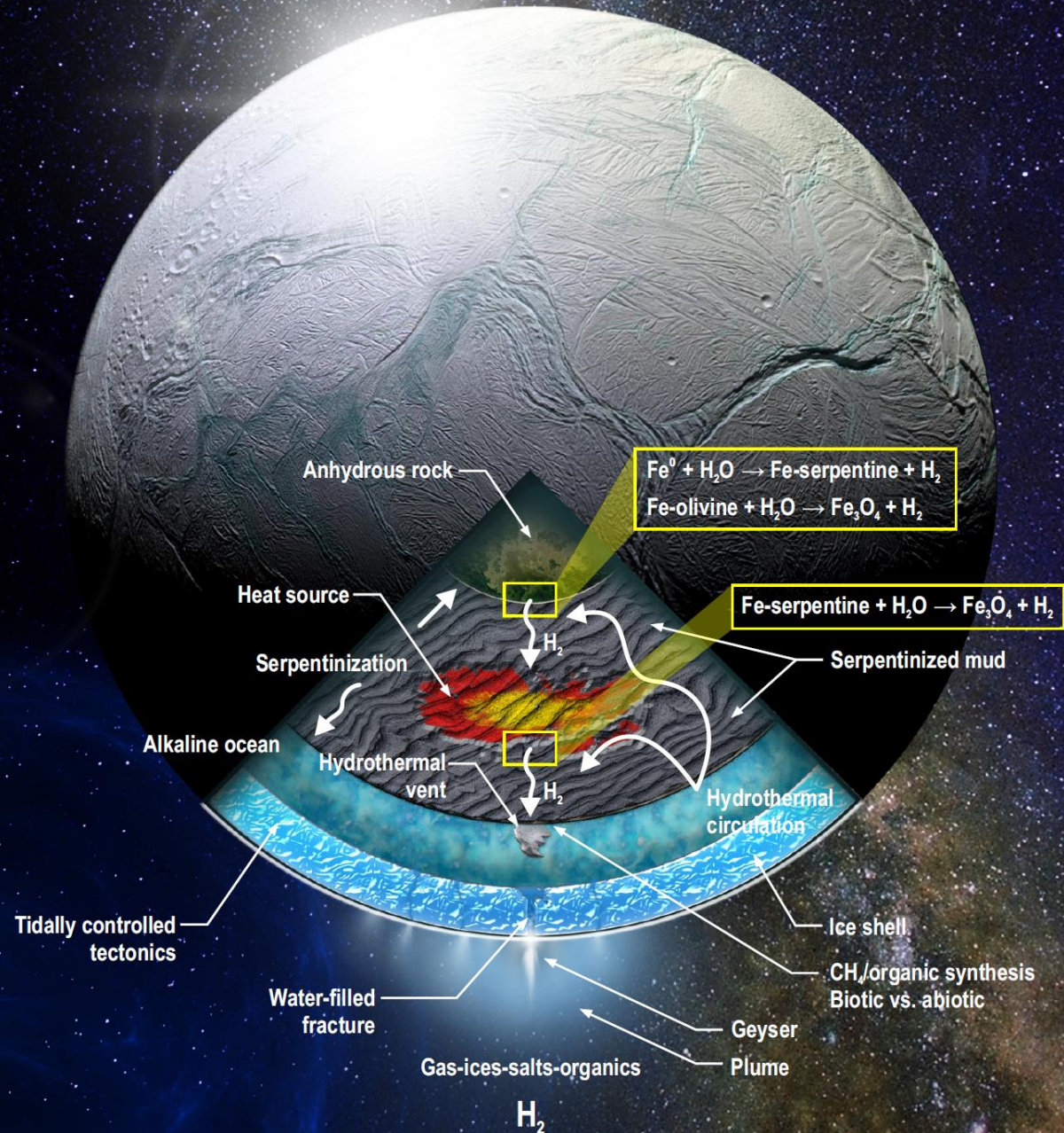


Parameter	Value
Temperature	90°C
pH	9-11
<b>H<sub>2</sub> conc.</b>	<b>10 mM</b>
CH <sub>4</sub> conc.	1 mM

Kelley et al. (2001; 2005), Proskurowski et al. (2006), Reeves et al. (2014)

mM = mmol per kg of H<sub>2</sub>O





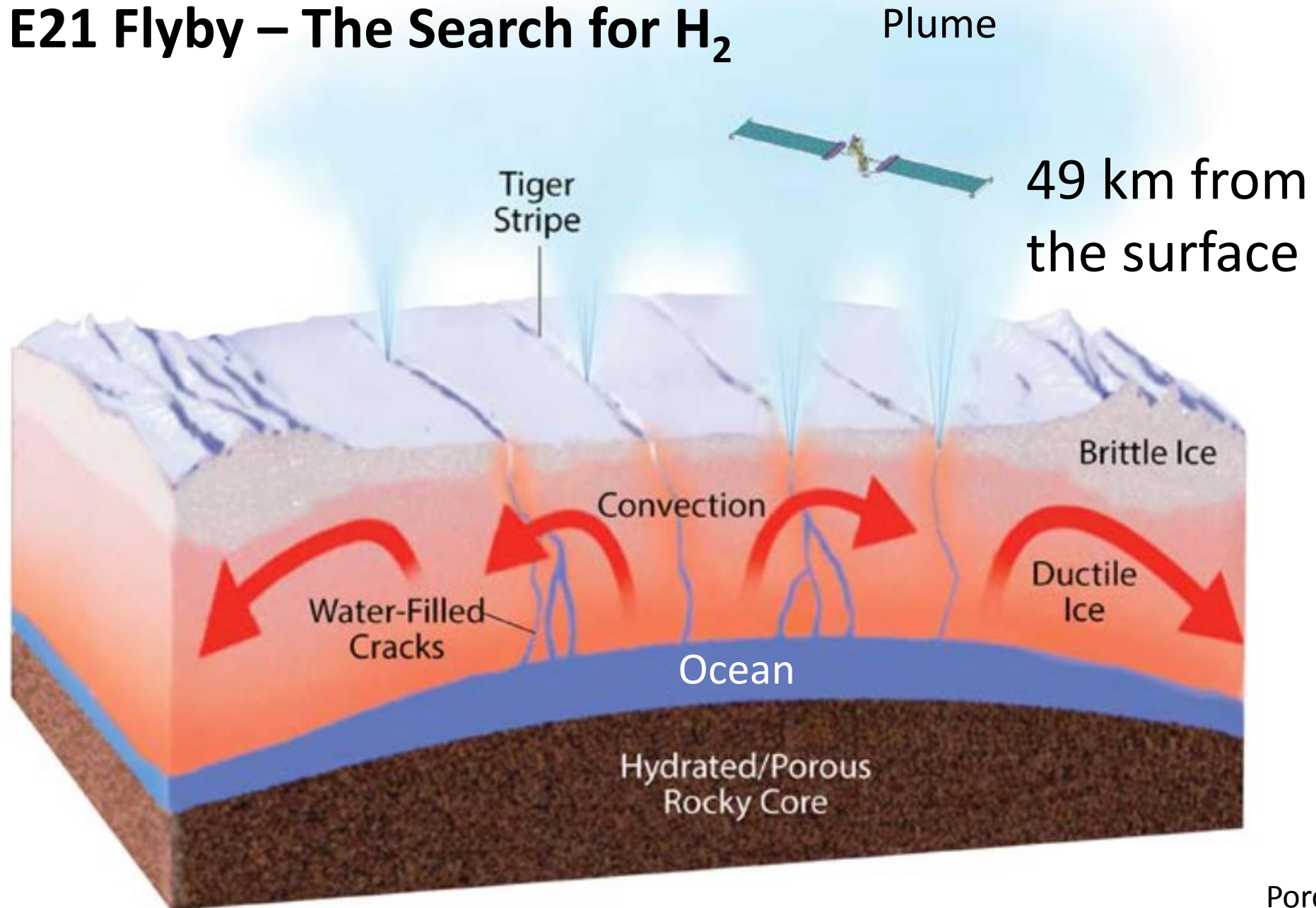
Model of  
**hydrothermal**  
**serpentinization**  
 suggested by  
 previous data

$\text{H}_2$  is the key  
 missing piece

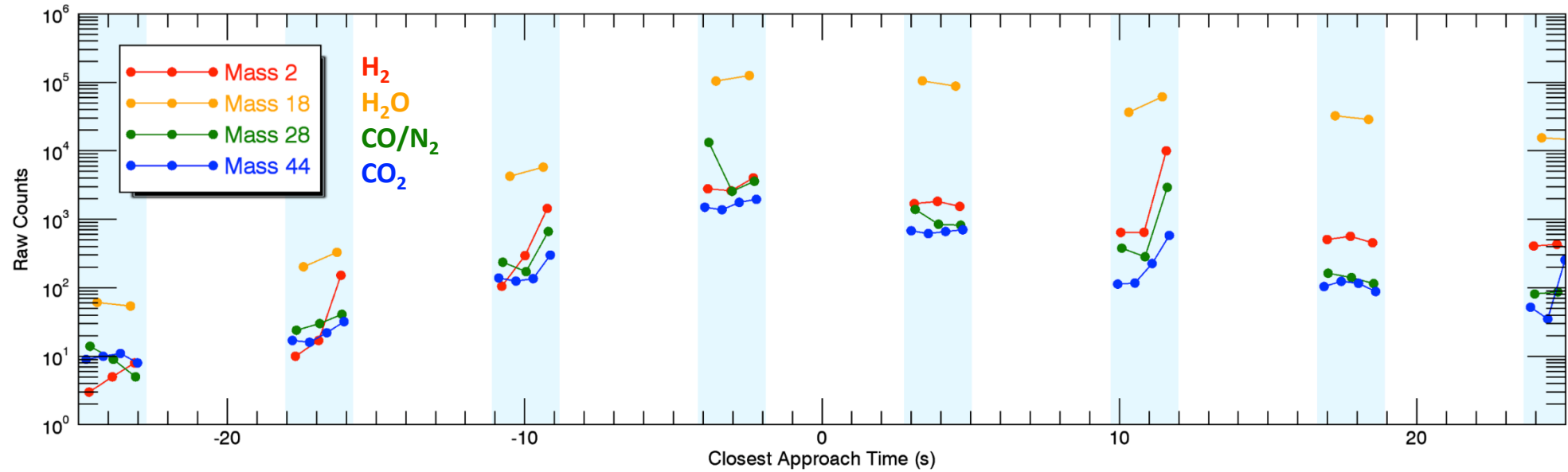


# Cassini

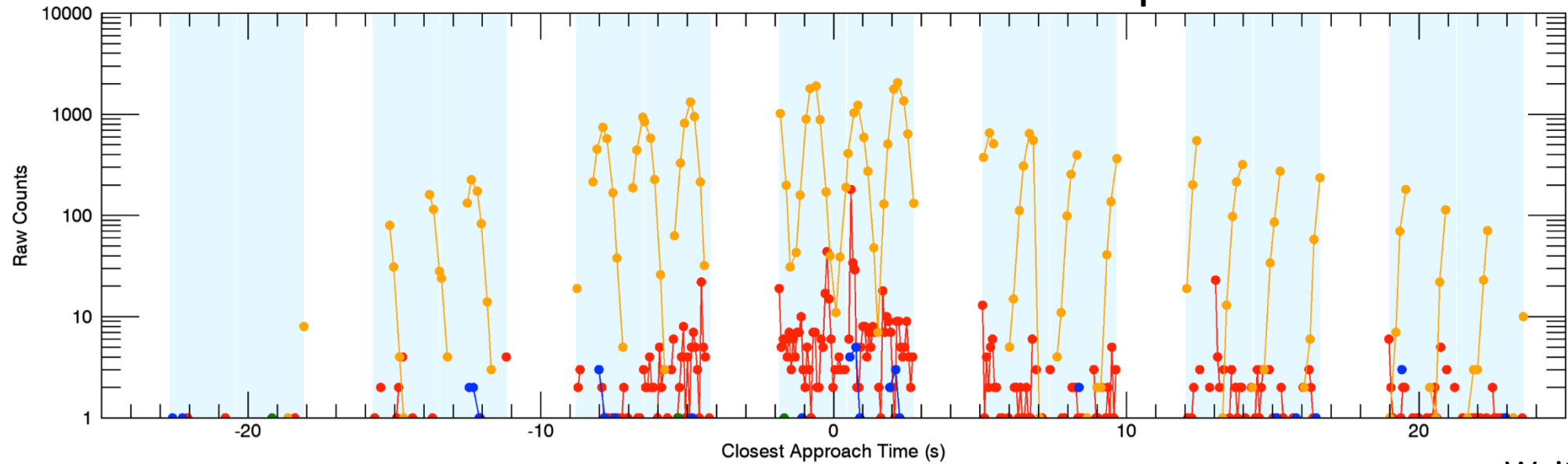
## E21 Flyby – The Search for H<sub>2</sub>



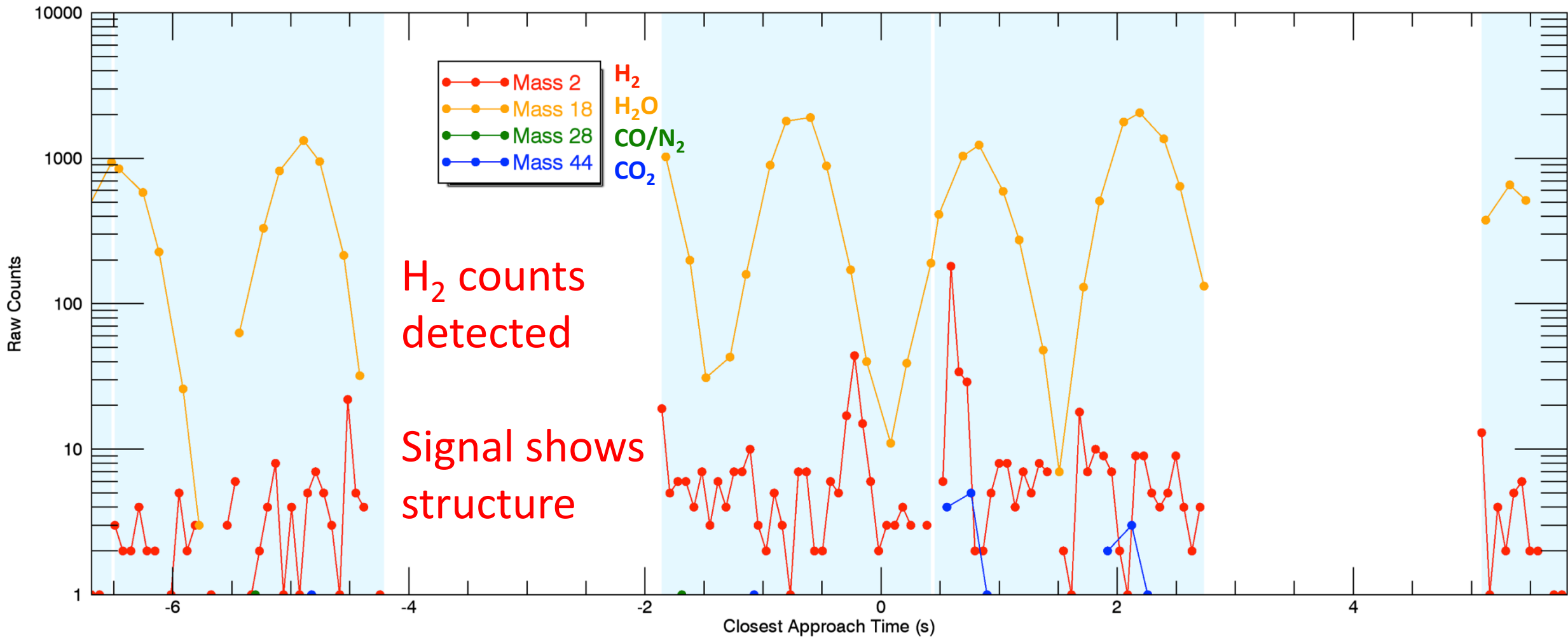
# Closed Source = Old



# Open Source = New







## Mass 2 Counts Observed in OSNB Mode

$H_2$  &  $H_2O$  enter the OPEN SOURCE

- ①  $H_2$  from Enceladus
- ②  $H_2^+$  generated from dissociative ionization of  $H_2O$  in the open source

Cross section obtained from Itakawa and Mason,  
J. Phys. Chem. Ref. Data, Volume 34, Number 1, 2005

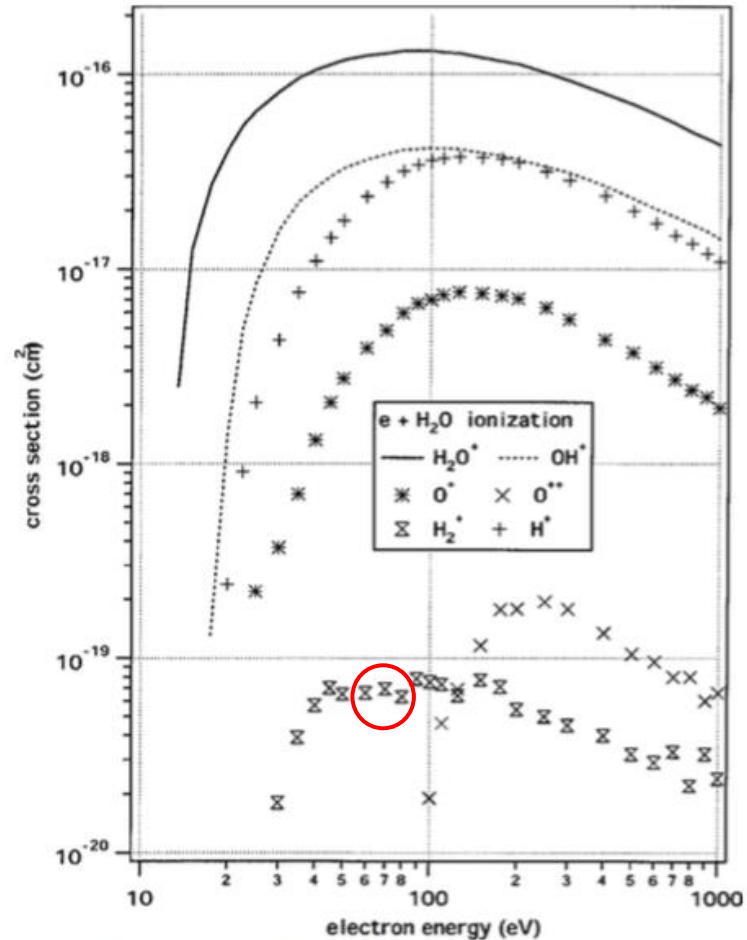
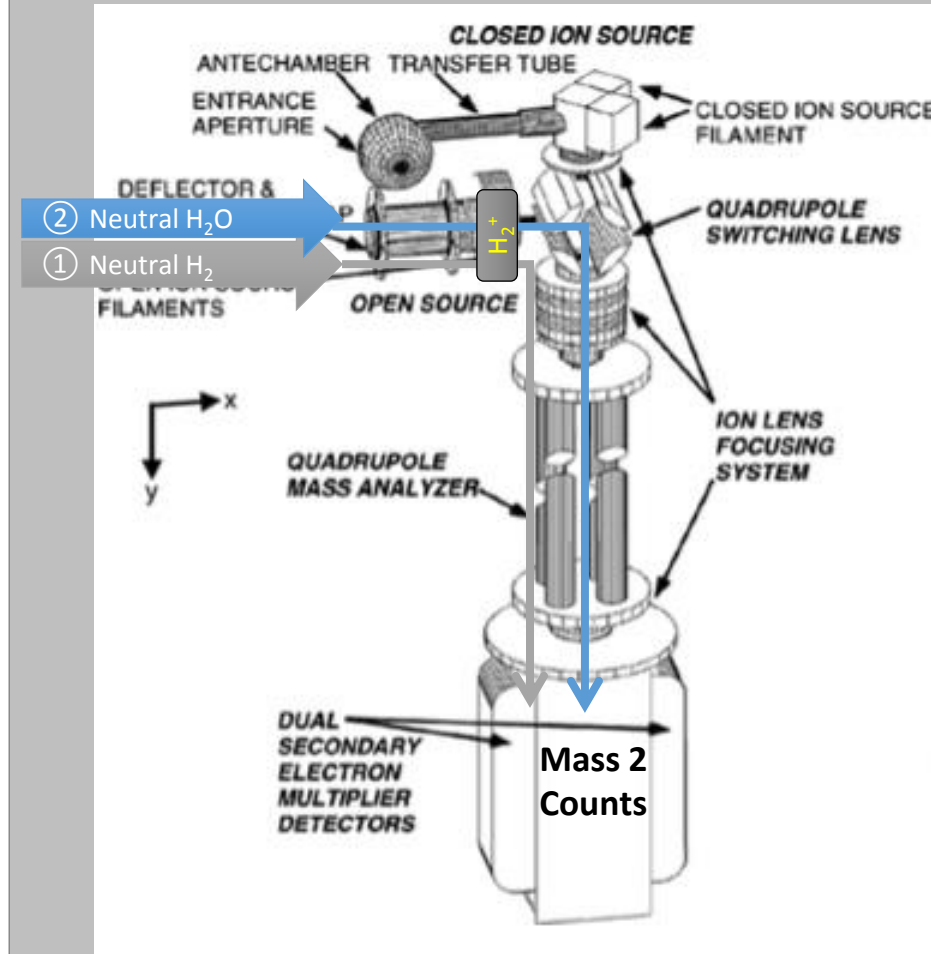


FIG. 11. Recommended values of the partial ionization cross sections of  $H_2O$  for the production of  $H_2O^+$ ,  $OH^+$ ,  $O^+$ ,  $O^{++}$ ,  $H_2^+$ , and  $H^+$ .

## Is the detected $H_2$ real?



## Instrument Background Estimation



## Mass 2 Counts Observed in OSNB Mode

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Cross section obtained from Itakawa and Mason, , J. Phys. Chem. Ref. Data, Volume 34, Number 1, 2005

$H_2$  &  $H_2O$  enter the CLOSED SOURCE

- ③  $H_2$  and  $H_2O$  gas ionized in the closed source and  $H_2^+$  leaks through the potential barrier on the quad lenses and into the quadrupole (Measured in INMS lab)
- ④  $H_2$  and  $H_2O$  gas (not yet ionized in the closed source) travels into the open source ionization region (Measured in INMS lab)

*Thermal gas in the instrument*

- ⑤ Thermal  $H_2$  &  $H_2O$  gas measured during OSNB mode (Measured in INMS lab)

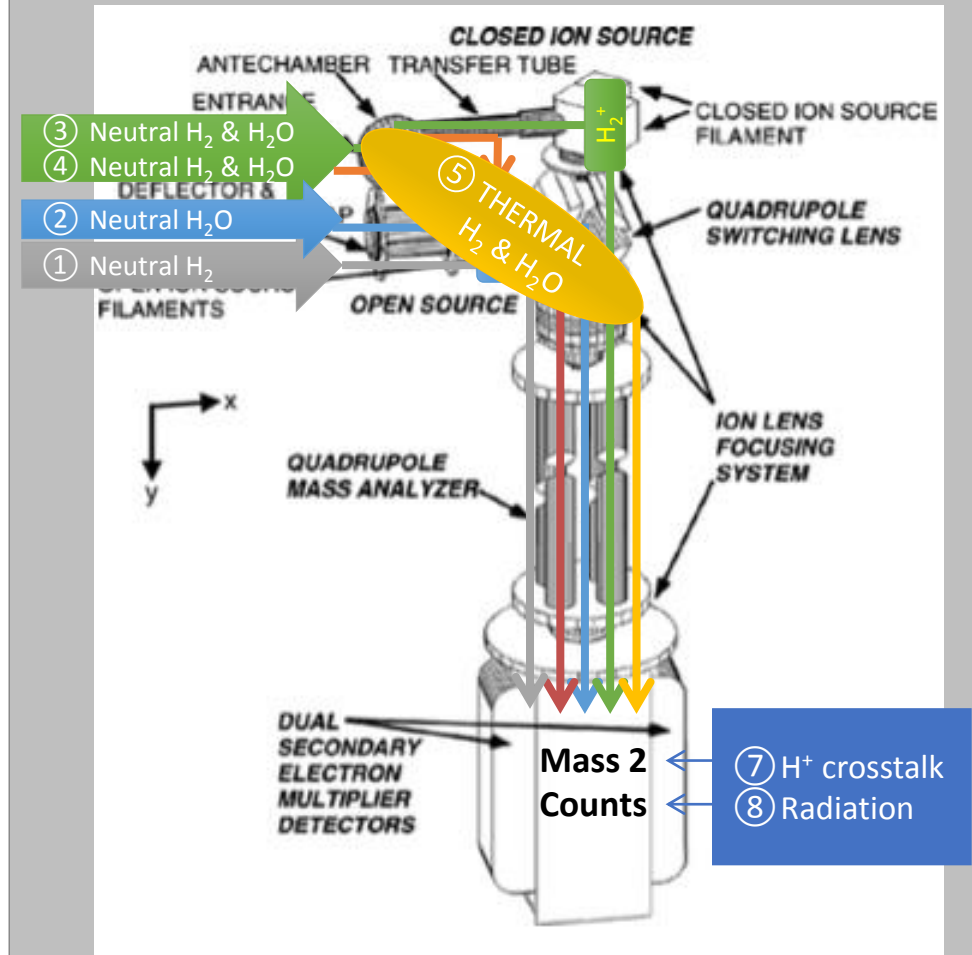
*INMS surface effects*

- ⑥  $H_2$  created from interactions with the surfaces of the instrument

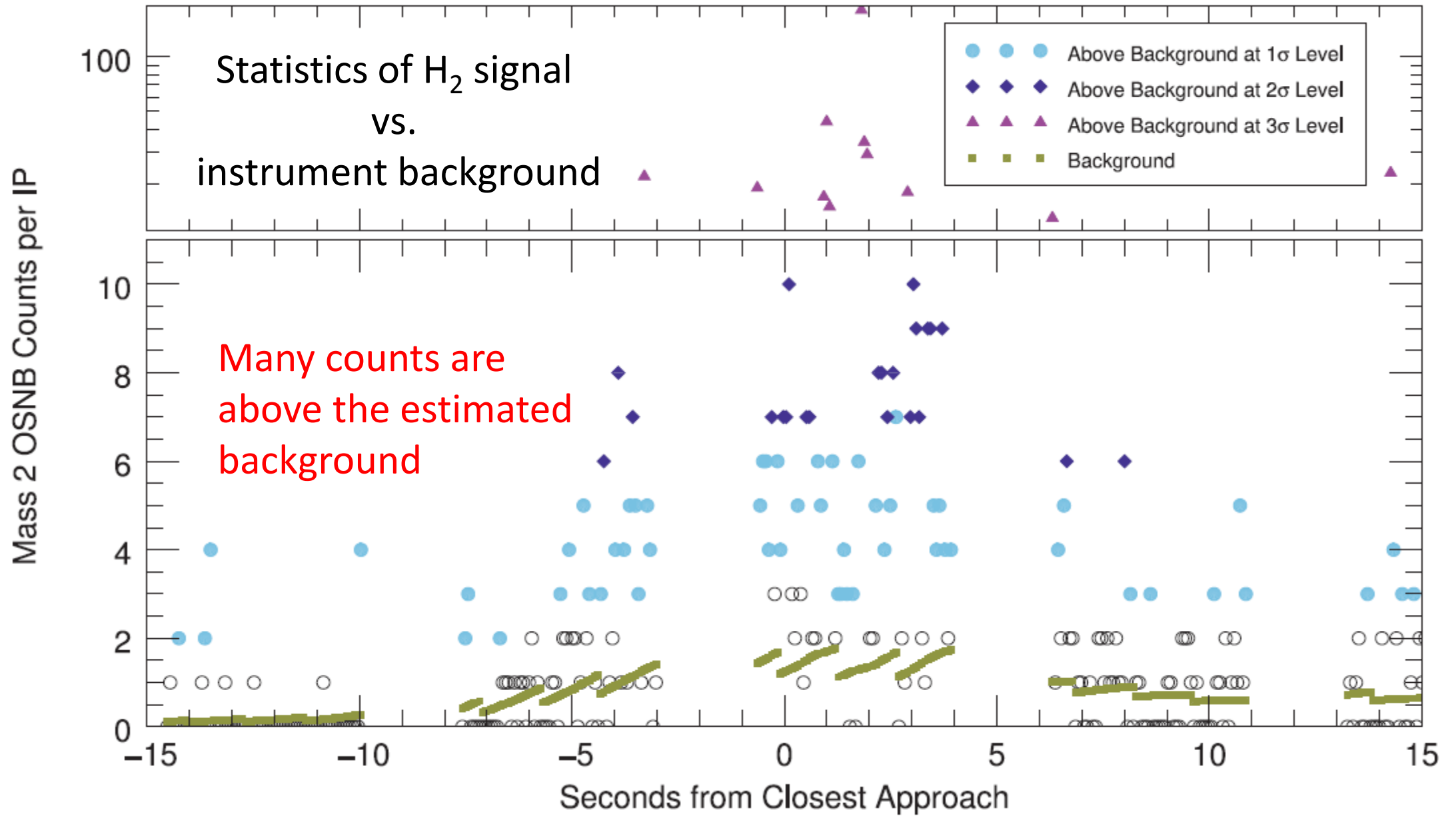
*Other sources of Mass 2 counts*

- ⑦  $H^+$  crosstalk
- ⑧ Radiation background

# Is the detected $H_2$ real?

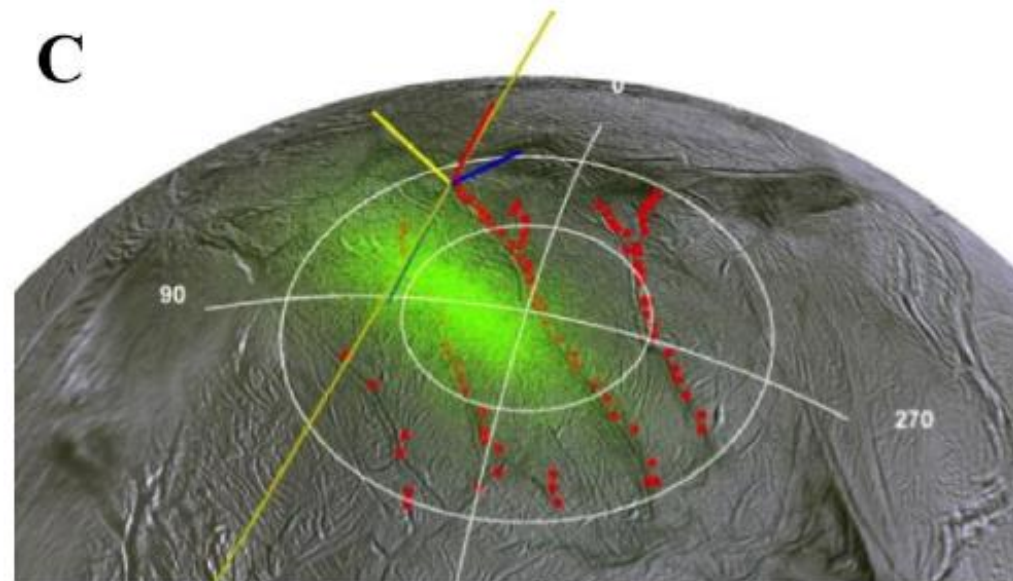
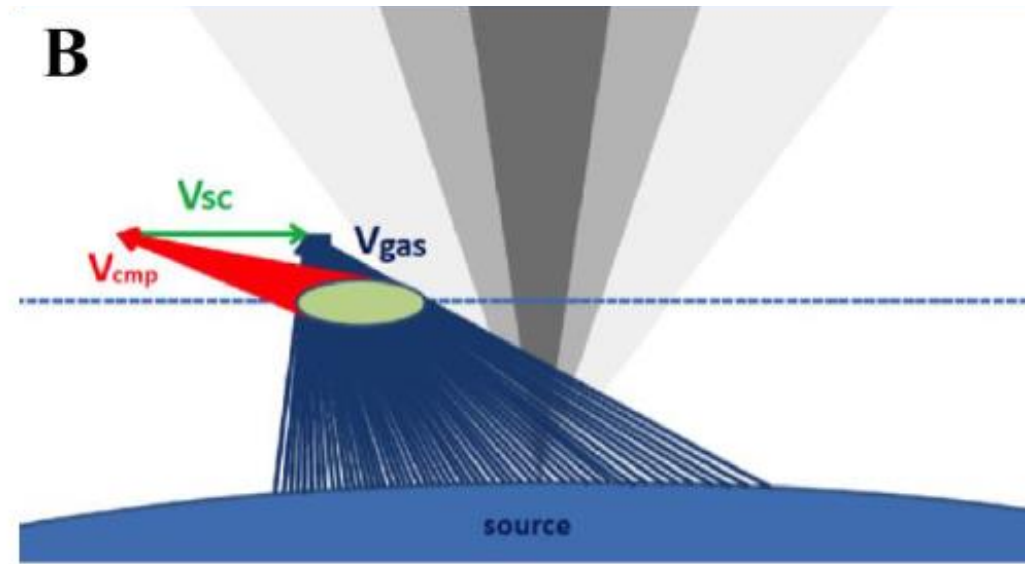


## Instrument Background Estimation

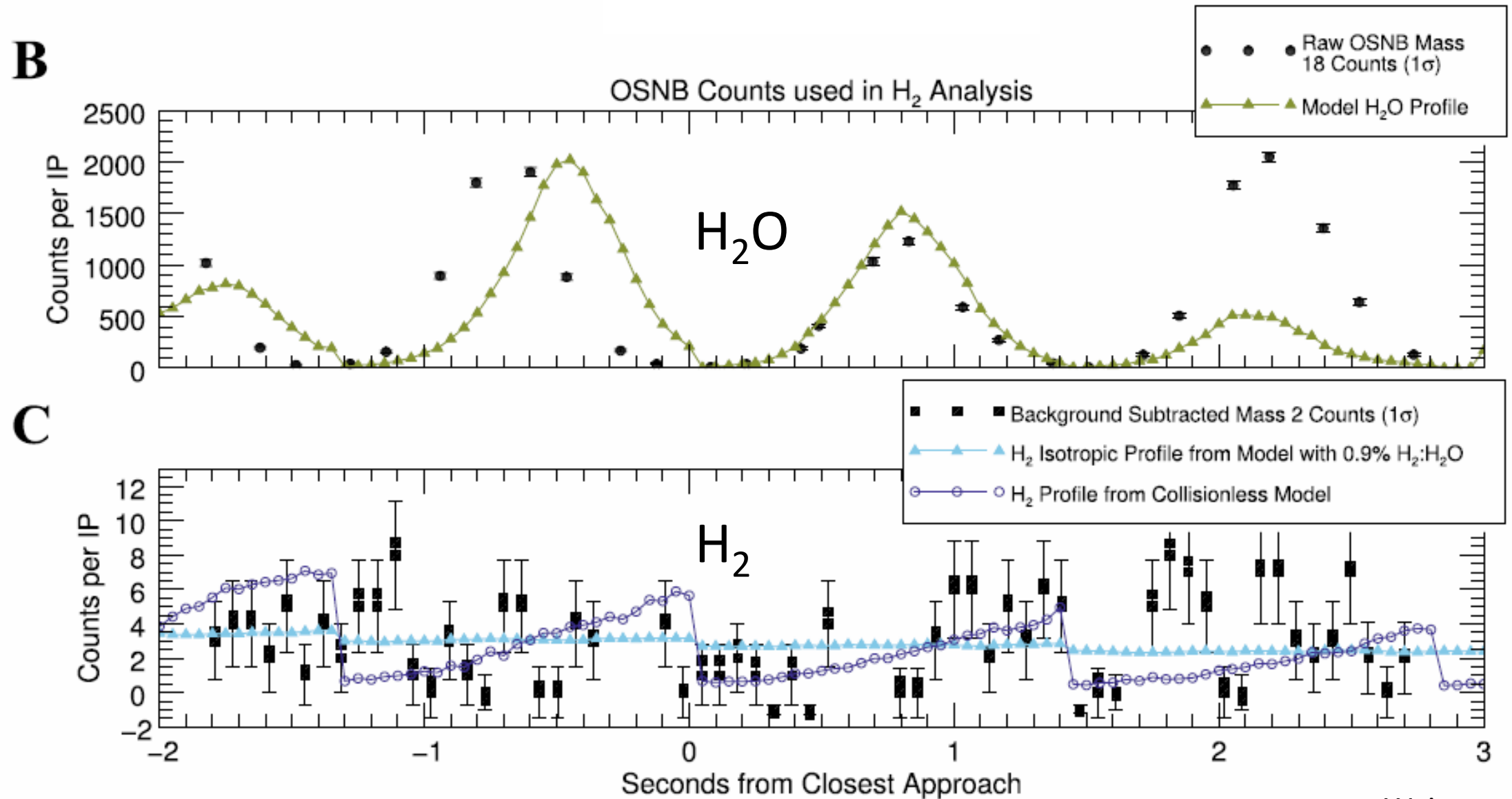




# Model of plume outflow to derive the H<sub>2</sub> fraction



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H<sub>2</sub> = Hydrothermal?





# H<sub>2</sub> = Hydrothermal?

Theoretical,  
Observational

Alternative source	Suggested inconsistency
Gravitational capture of nebular H <sub>2</sub> e.g., Pollack et al. (1996)	Enceladus too small (>10M <sub>E</sub> ), He not detected in plume
Trapping of H <sub>2</sub> in amorphous ice (<20 K) e.g., Bar-Nun & Prialnik (1988)	No evidence of such cold material in comets (OPR), lack of Ar, Ne, CO/N <sub>2</sub> in plume

Primordial

Homemade

# H<sub>2</sub> = Hydrothermal?

Theoretical,  
Observational

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Homemade	Radiolysis of water ice on surface e.g., Cooper et al. (2009)	Would not be concentrated in plume, low radiation fluxes at Enceladus, O <sub>2</sub> not detected in plume
	Radiolysis of liquid water in interior e.g., Chyba & Hand (2001)	Low Cl chondritic radionuclide abundances, H <sub>2</sub> /CH <sub>4</sub> ratio too high in plume

# A Bottom-Up Test of the Hydrothermal Model

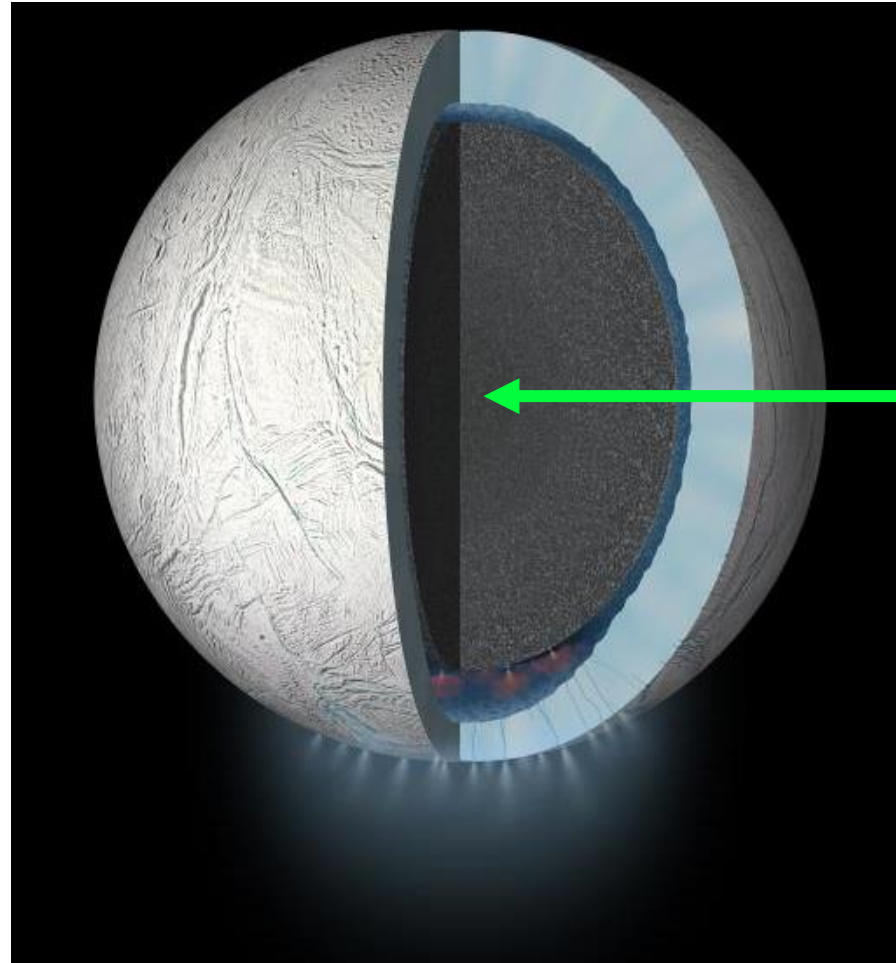
- *Main idea:  $H_2$  production from  $H_2O$  is coupled to Fe oxidation*
- As in hydrothermal systems on Earth because of the high abundance of Fe
- Key geochemical reactions in the Fe-Si-O-H system:
  - a.  $3Fe^0 + 5H_2O + 2SiO_2 \rightarrow \text{Fe-serpentine} + 3H_2$
  - b.  $3\text{Fe-olivine} + 2H_2O \rightarrow 2Fe_3O_4 + 3SiO_2 + 2H_2$
  - c.  $\text{Fe-serpentine} \rightarrow Fe_3O_4 + H_2O + 2SiO_2 + H_2$
- Approach: Estimate  $H_2$  yield from amounts of Fe minerals on Enceladus





# A Bottom-Up Test of the Hydrothermal Model

- Mass of rock from the internal structure model of McKinnon (2015)

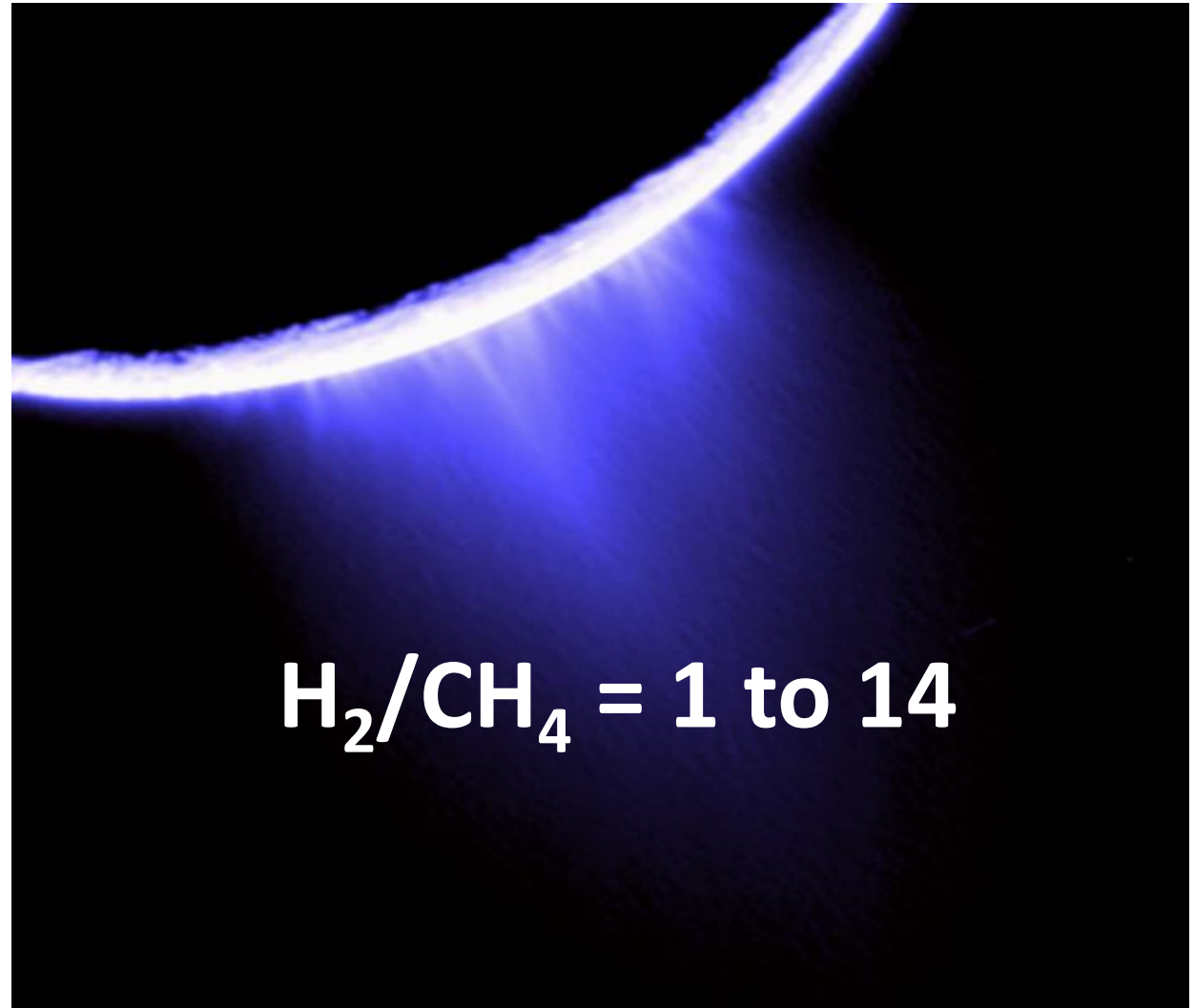
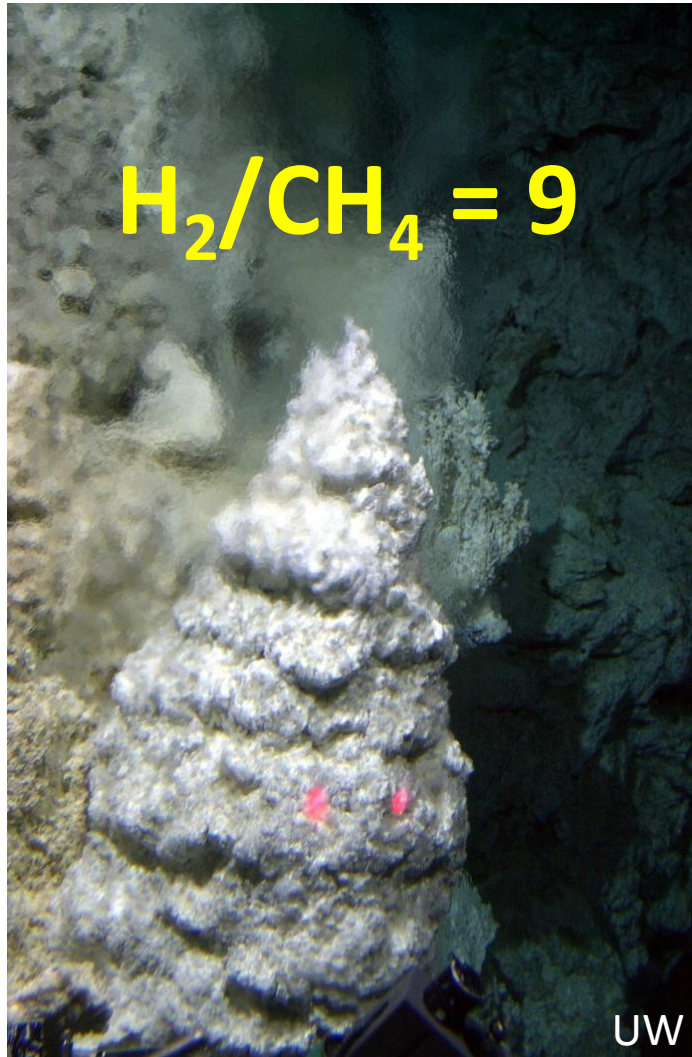


Rock = Source of  
electrons to make  
 $H_2$  from  $H_2O$

# A Bottom-Up Test of the Hydrothermal Model

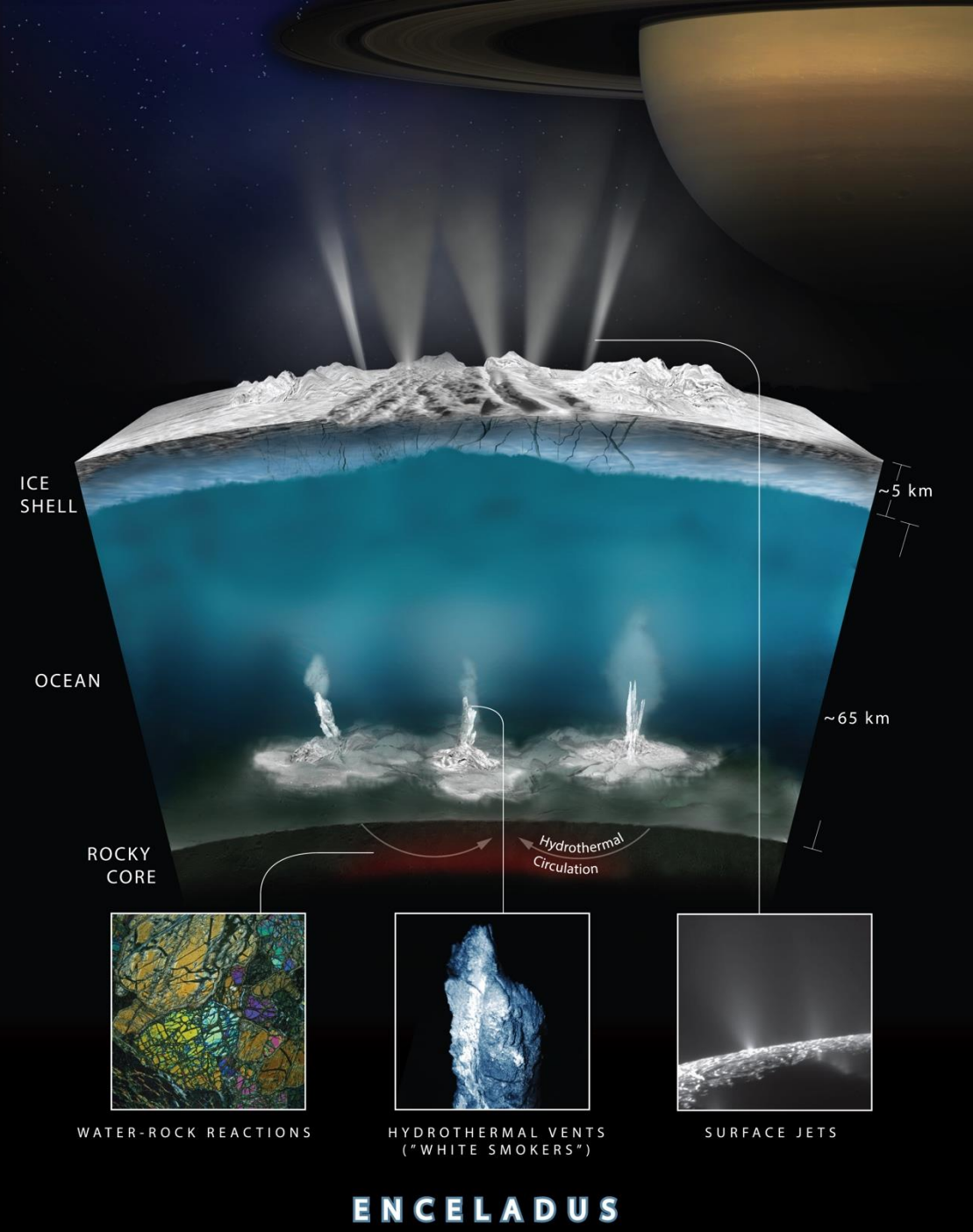
- Mass of rock from the internal structure model of McKinnon (2015)
  - Mineralogy of rock based on solar elemental abundances (Lodders, 2003) and alteration phases in carbonaceous chondrites (Brearley, 2006)
- 
- **Example:** 1% anhydrous accreted rock in the core can sustain  $\sim 1\%$   $\text{H}_2$  in the plume at today's outgassing rate (Hansen et al., 2011) for  $\sim 500$  Myr
  - The presence of appreciable  $\text{H}_2$  in the plume does not require a large amount of anhydrous rock. Less if outgassing is only episodic
  - Compatible with a low density core (McKinnon, 2015) that may be dominated by hydrated silicates containing some pore water

$H_2/CH_4$  ratio similar to Lost City vents





The hydrothermal model is supported by the data



<https://photojournal.jpl.nasa.gov/catalog/PIA21442>

# Comparative Deep Energy Earth vs. Enceladus



Body:	Earth	Enceladus	Enceladus/Earth
Production Rate:	$1 \times 10^{12}$ mol H <sub>2</sub> /yr (Sherwood Lollar et al., 2014)	$3 \times 10^9$ mol H <sub>2</sub> /yr (~1% H <sub>2</sub> in plume)	0.003
Surface Flux:	2000 mol H <sub>2</sub> /yr km <sup>2</sup>	4000 mol H <sub>2</sub> /yr km <sup>2</sup>	2



# H<sub>2</sub> links the inorganic and organic/living worlds

- **Organic synthesis**

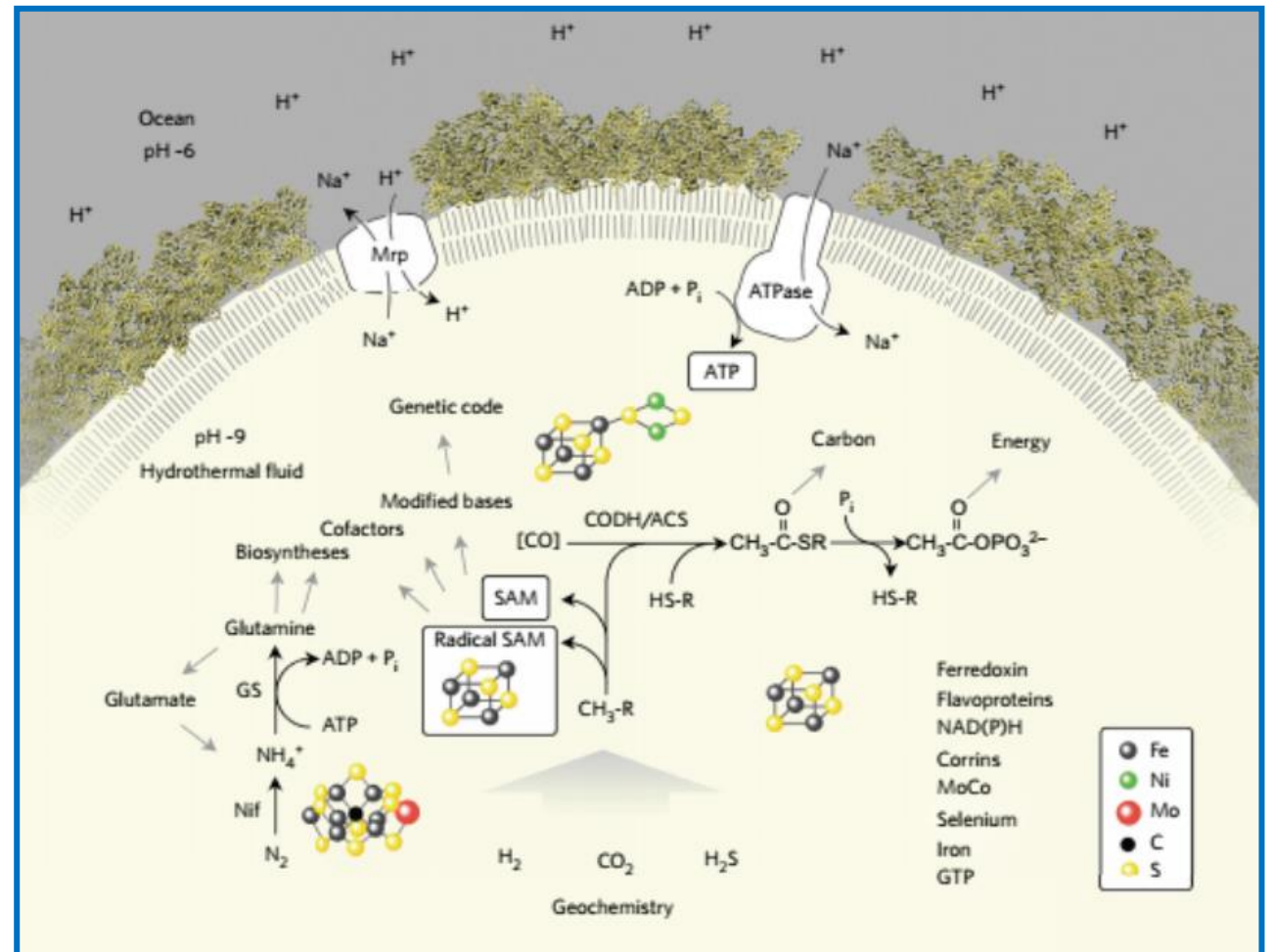


- **Prebiotic chemistry**

Current model: Life began at *alkaline* hydrothermal vent

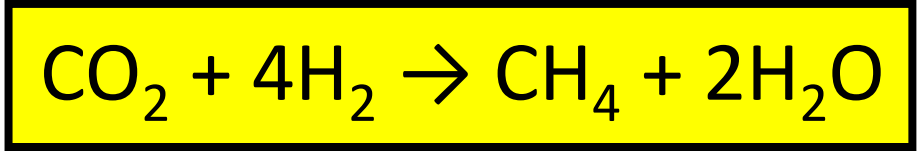
- **Chemical energy for life**

H<sub>2</sub>/CH<sub>4</sub>-based metabolisms



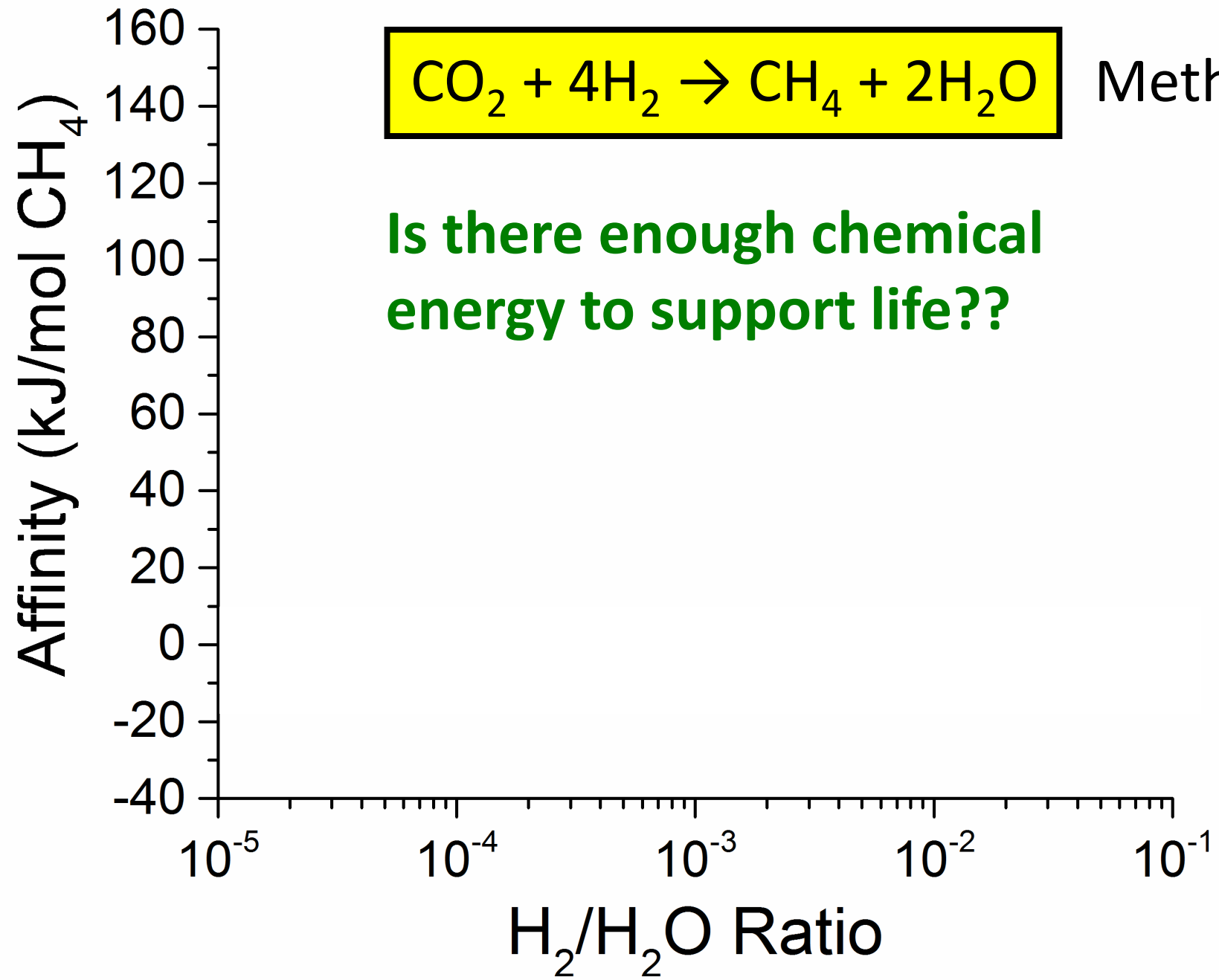
Weiss et al. (2016, Nat. Microbiol. 1, 16116)

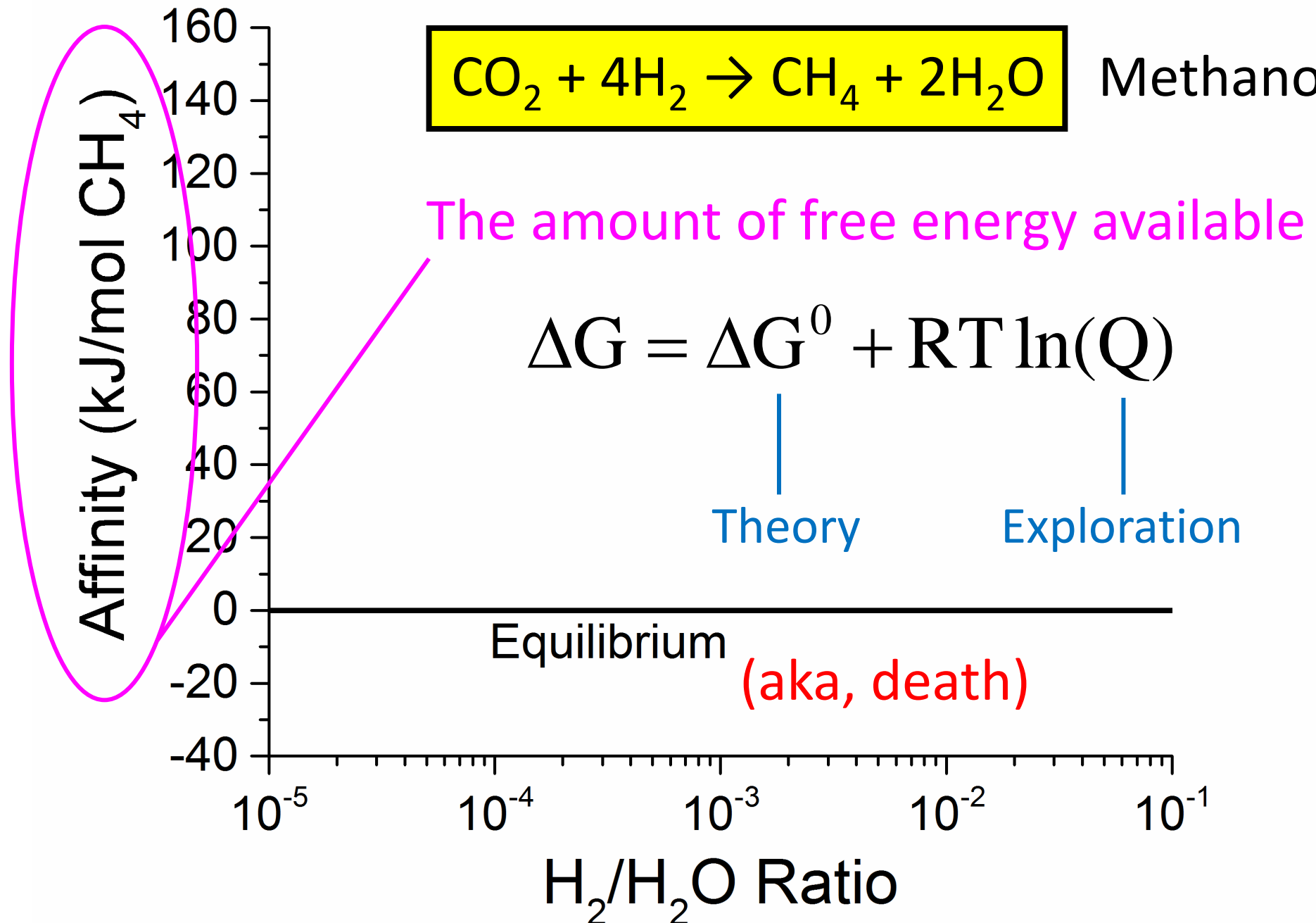


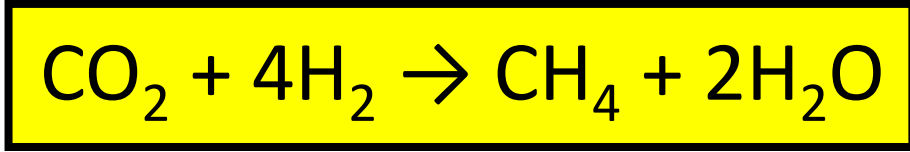


Methanogenesis

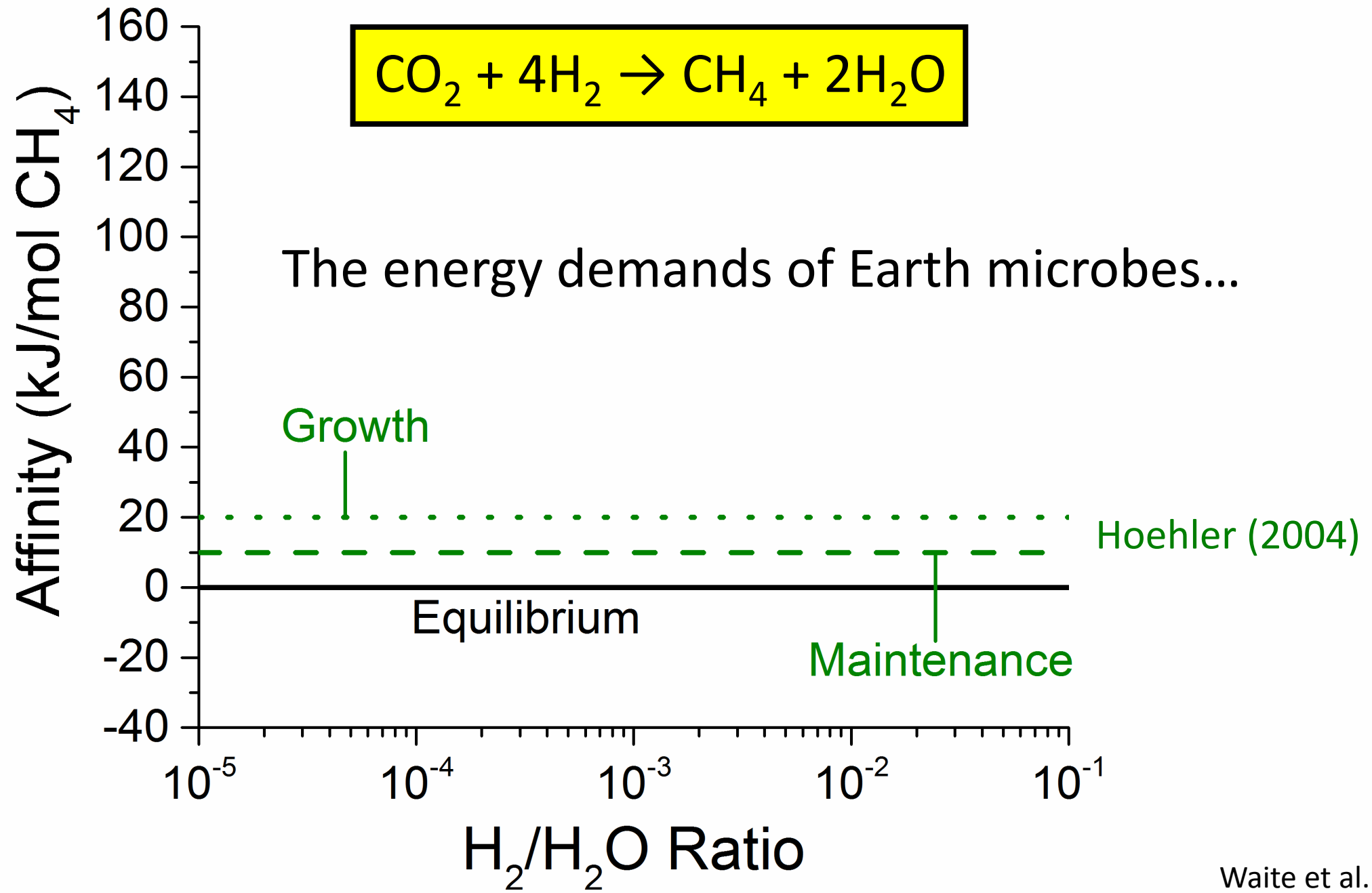
Is there enough chemical energy to support life??

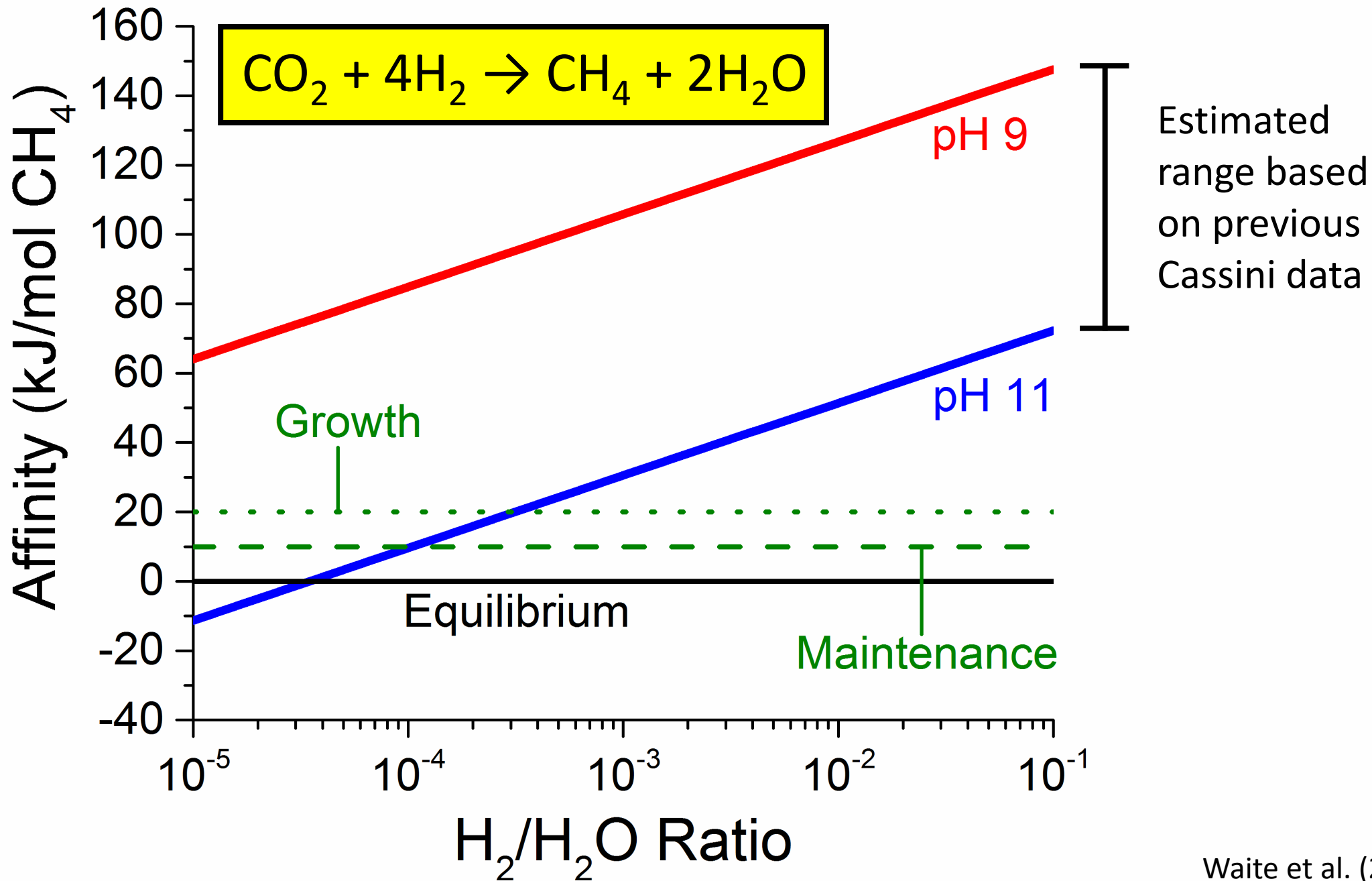




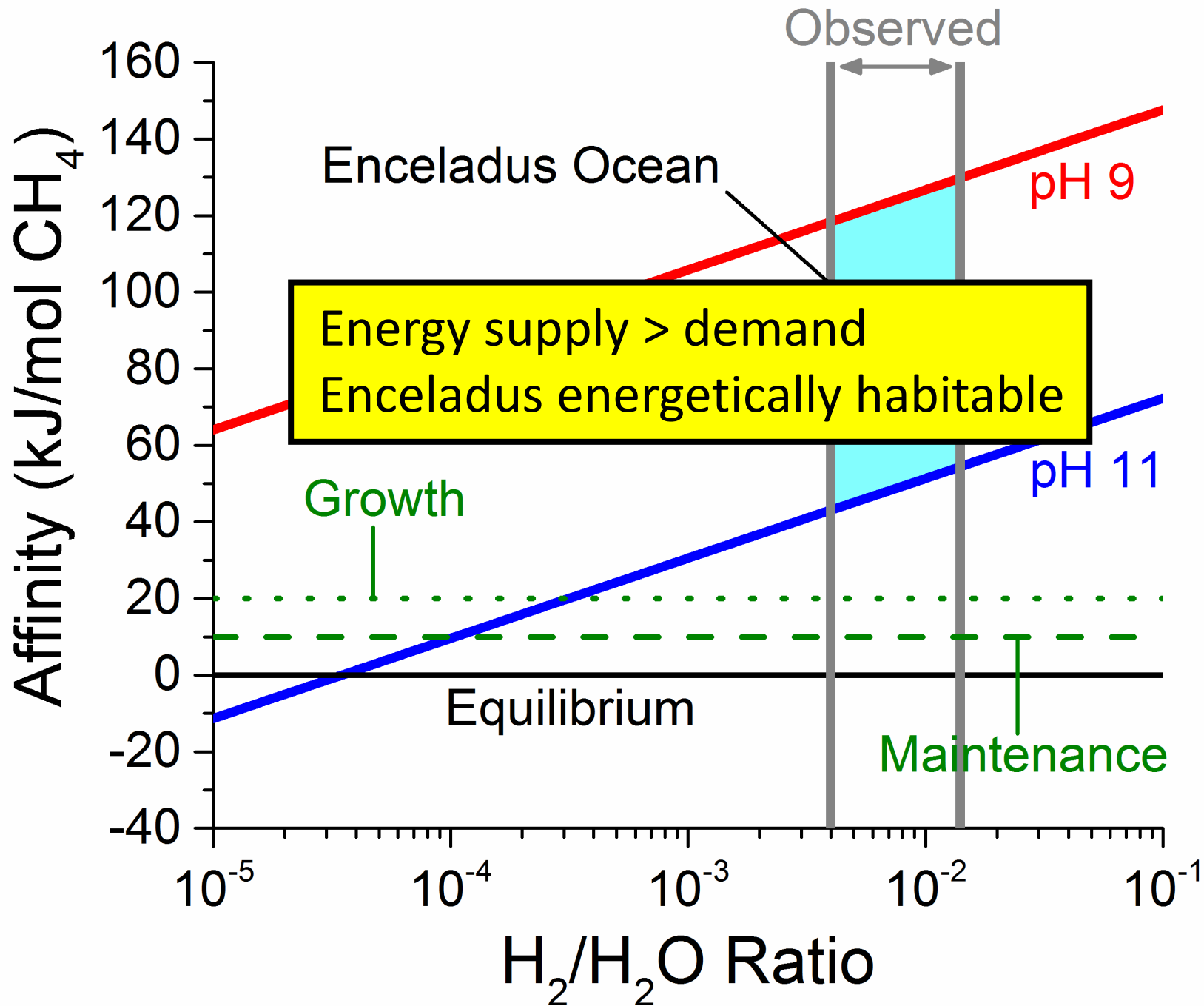


The energy demands of Earth microbes...









Is there enough chemical energy to support life??

YES!



# H<sub>2</sub> evidence that Enceladus is energetically habitable



Jet Propulsion Laboratory  
California Institute of Technology

Education | Intern Learn Teach News Events



Habitability: ✓

HOT-N-READY  
CHEESE



Caloric equivalent of  
of ~300 pizzas  
every hour!

Free  
Energy

Liquid  
Water

Essential  
Elements ✓

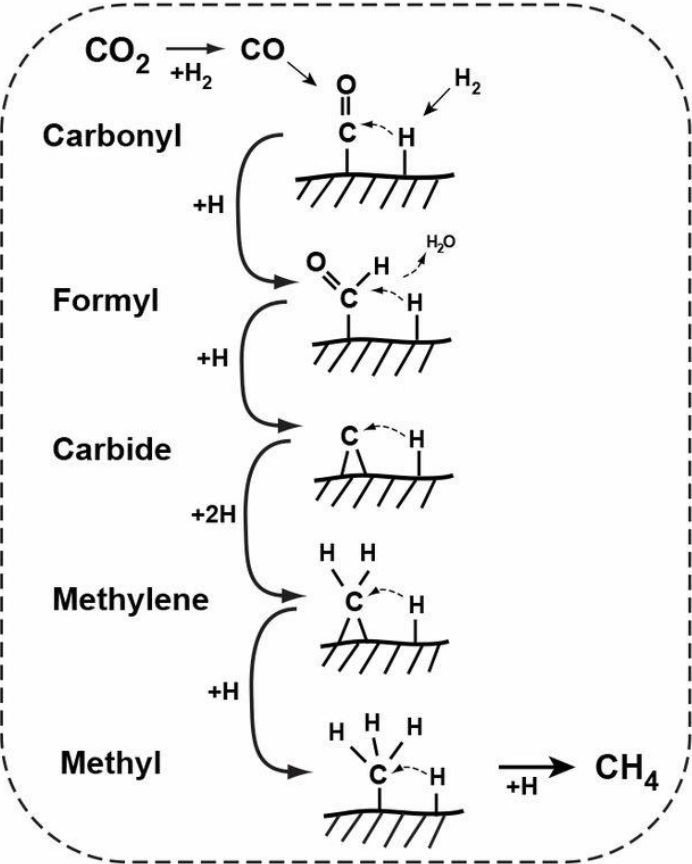
C, N, O, S? P?

Inhabited?

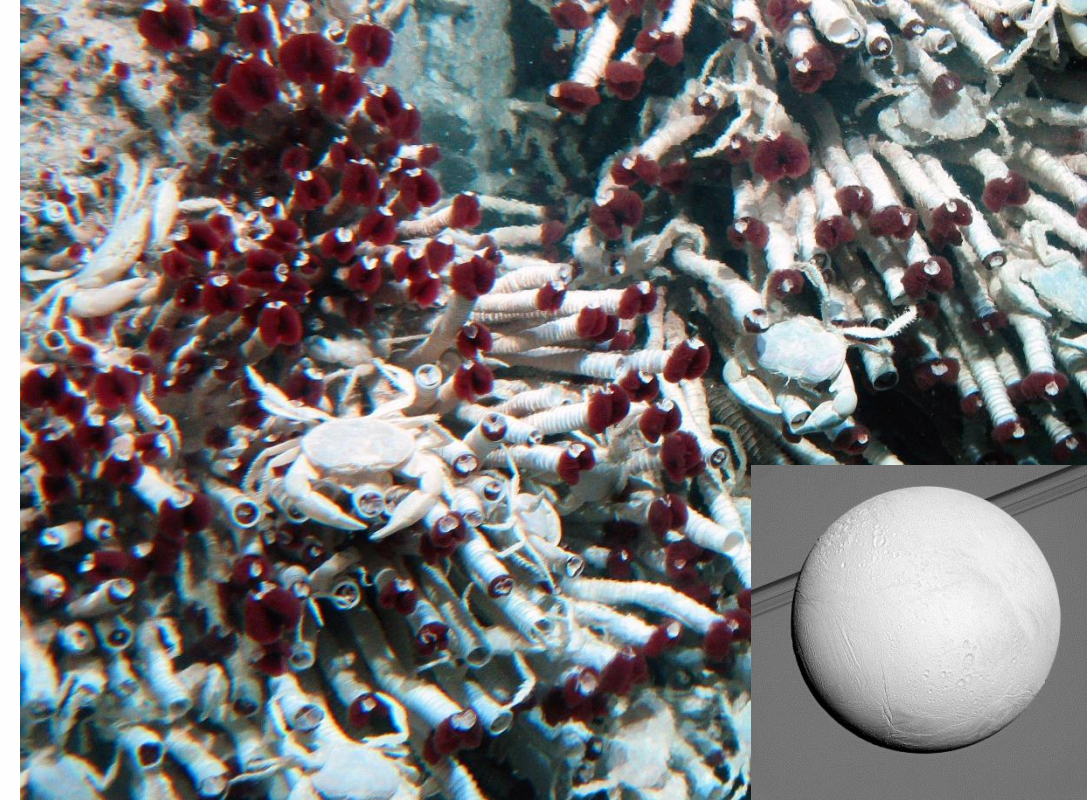
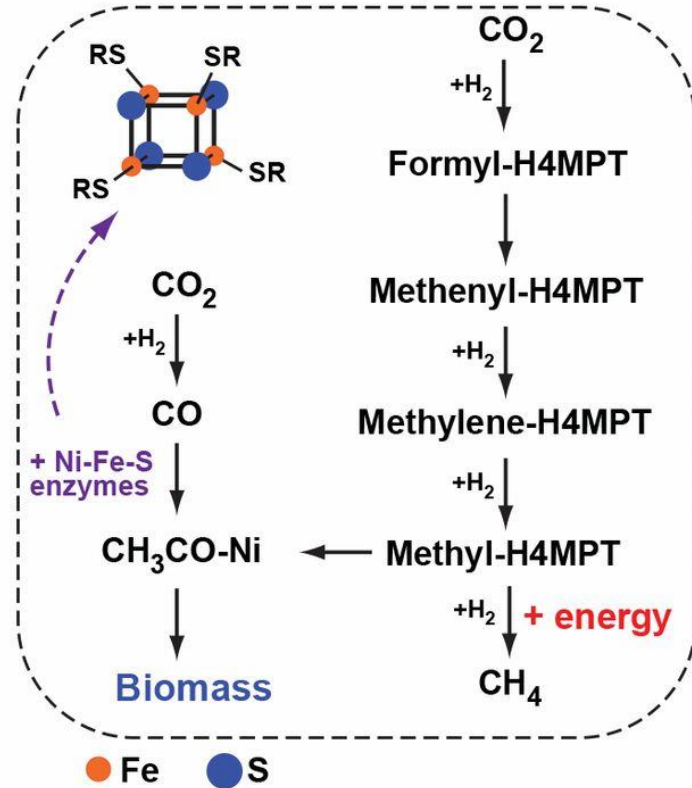
That's the **\$1B**  
question!

# H<sub>2</sub> and the drive to life as the next theme of Enceladus exploration

Abiotic pathway:



Biotic pathway:



McCollom & Seewald (2013)

Geochemistry → Biochemistry → Ecology?



# Energy availability sets the stage for future missions exploring relationships between habitability and the presence of life on ocean worlds



The Law of Free Food:

$$\text{Food Taste} = \frac{\text{Food Quality} \times \text{Hunger}}{\text{\$ Cost}}$$





# Conclusions

- Geochemical model of hydrothermal serpentinization
- Identified native H<sub>2</sub> in the plume from INMS
- Hydrothermal source of H<sub>2</sub>
- H<sub>2</sub> is a potent and ancient energy source for microbes
- Made the first calorie count of an alien ocean
- Follow the H<sub>2</sub>!

